$\qquad$
How many calories are in a peanut?
If we put a small beaker close to the flame of a burning peanut, and heat a small amount of water, say 100 g , we can collect about $30 \%-50 \%$ of the heat from the flame.

It takes 4180 J to raise the temperature of 1 kg of water 1 degree.
Finding the mass of half a peanut is a little tricky. Maybe you could measure the mass of 20 halves.

One serving is 28 grams. One Calorie is 1000 Joules. How many calories will this provide for a person? $\qquad$ Check this against published info on the web.

Show your data and work
Initial temperature of water $\qquad$
Final temperature of water $\qquad$
Mass of water $\qquad$
Mass of peanut burnt $\qquad$
Joules of heat absorbed by water $\qquad$
Calories in 28 g of peanuts. $\qquad$

